



# Interpretation of Mixed Acid Base Disorders- No Da Vinci Code

by

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# ARTERIAL BLOOD GASES

- Measured on a Blood Gas Analyser.
- Sample of whole blood is directly injected into the analyser without delay



# Site for Sample Collection

- Best site ?



- Why ?
- No vein adjacent to Radial Artery

# Arterial or Venous Blood ??

	arterial blood	venous blood	COPD
pH	7.35-7.45	7.31-7.40	7.27
PCO <sub>2</sub>	35-45mmHg (4.7-6.0 kPa)	41-51mmHg (5.5-6.8kPa)	54mmHg
HCO <sub>3</sub> <sup>-</sup>	22-28mmol/l	25-29mmol/l	31mmol/l
PO <sub>2</sub>	> 80mmHg (>10.6kPa)	38-42 mmHg (5.1-5.6kPa)	40mmHg



**Solution ?**

**Mention site on Lab Form**

# Measured and Calculated Tests

## Measured parameters:

- pH

- *Partial Pressure of Carbon dioxide ( $PCO_2$ )*

- *Partial Pressure of oxygen ( $PO_2$ )*

## • Calculated parameters:

- $HCO_3$  using Henderson Hasselbalch equation

- Base Excess (BE)

# ABG measurement – Some unique things

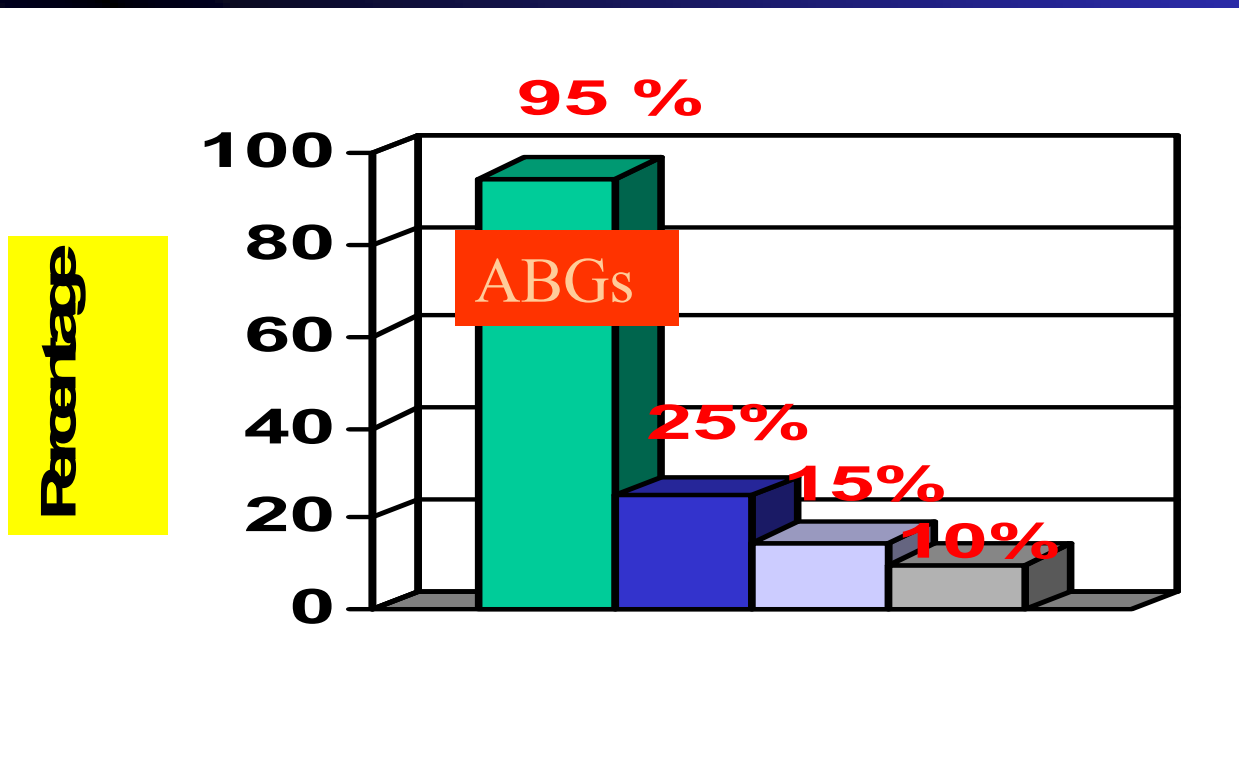
- Samples to reach lab within 15 minutes
- No incubation or reagent mixing
- Instant results
- Invalid if result delayed
- “bedside use” with some precautions

# ABG measurement – An expensive test

- Quite expensive investigation
- Under-utilized
- Cost of even one test per day is also very high
- Proper interpretation is required for optimum use



# ABG measurement – High diagnostic yield



 **ABGs**

 **CBC**

 **Electrolytes**

 **Urinalysis**

# Types of Acid Base Disorders

- Single (simple) disorders
- Double (mixed) disorders
- Triple (mixed) disorders



# Simple (Single) Disorders

- Metabolic Acidosis
- Metabolic Alkalosis
- Respiratory Acidosis
- Respiratory Alkalosis

# Double Disorders

- Metabolic and Respiratory Acidosis
- Metabolic and Respiratory Alkalosis
- Metabolic Alkalosis and Respiratory Acidosis
- Metabolic Acidosis and Metabolic Alkalosis



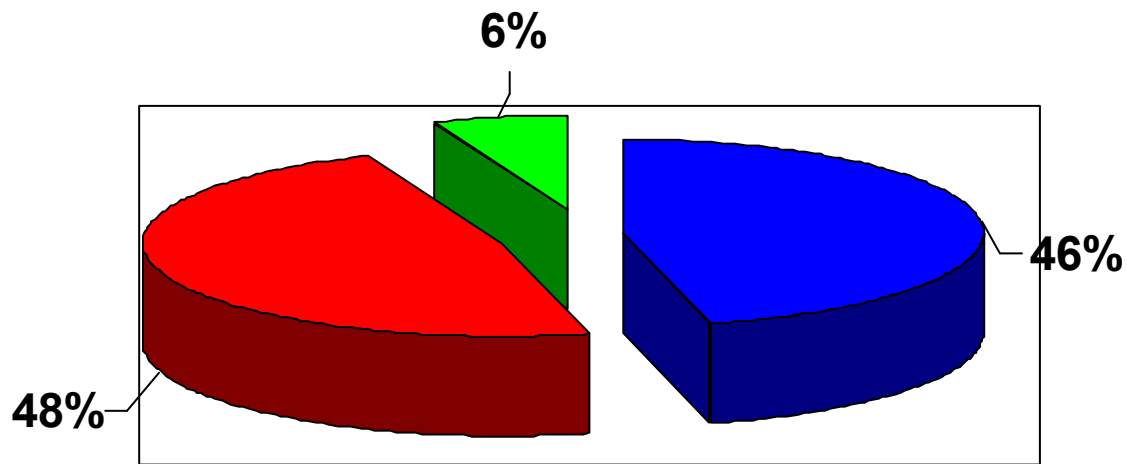
# Triple Disorders

Metabolic Acidosis, Metabolic Alkalosis  
and Respiratory Alkalosis:

beyond “salicylate poisoning”

*Any high anion gap acidosis  
complicated by vomiting and  
hyperventilation can cause this  
disorder*

# Pattern of ABG results from CCU,ICU and NICU in 45 days (n=114)

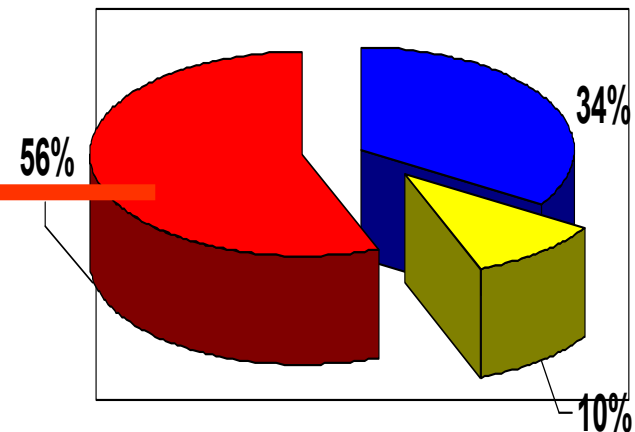
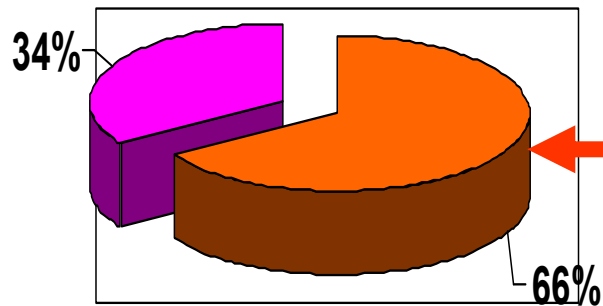


■ Simple Disorders

■ Mixed Disorders

■ Normal Results

# Distribution of Mixed Acid Base Disorders in a Critical Care Units (PNS SHIFA) in 45 days (n=67)



■ Metabolic Acidosis+Respiratory Alaklosis  
■ Metabolic Alkalosis +Respiratory Acidosis

■ Double Acidosis ■ Double Alkalosis ■ Opposing disorders

# Analysing ABG Reports

- Clinical features
- Results of other investigations e.g. electrolytes etc
- ABG result necessary for confirmation of diagnosis
- Biochemical opinion may be given even with meagre information



# Treatment Principles of Acid-Base Disturbances



## The 3-Cs

### Correct analysis

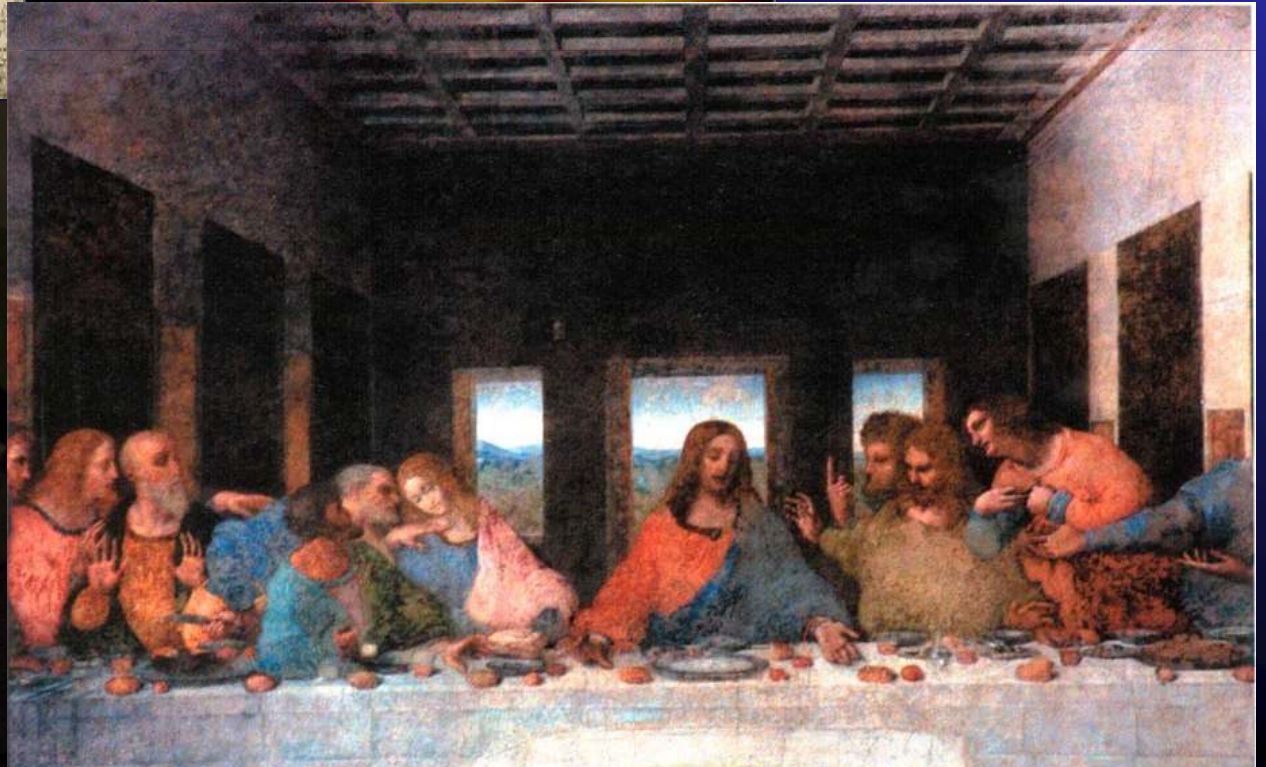
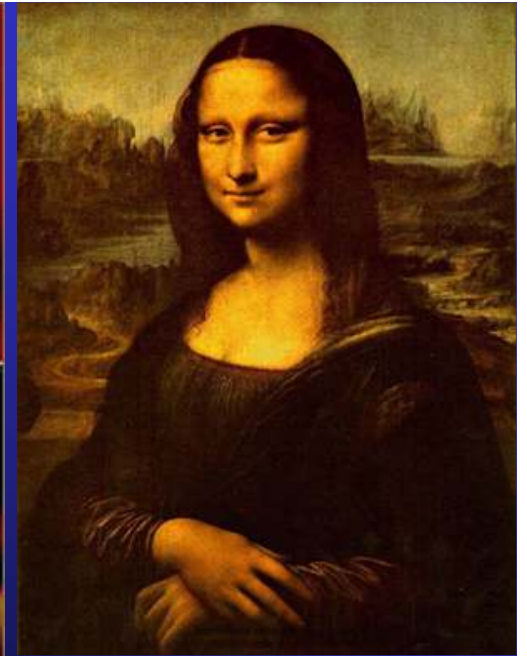
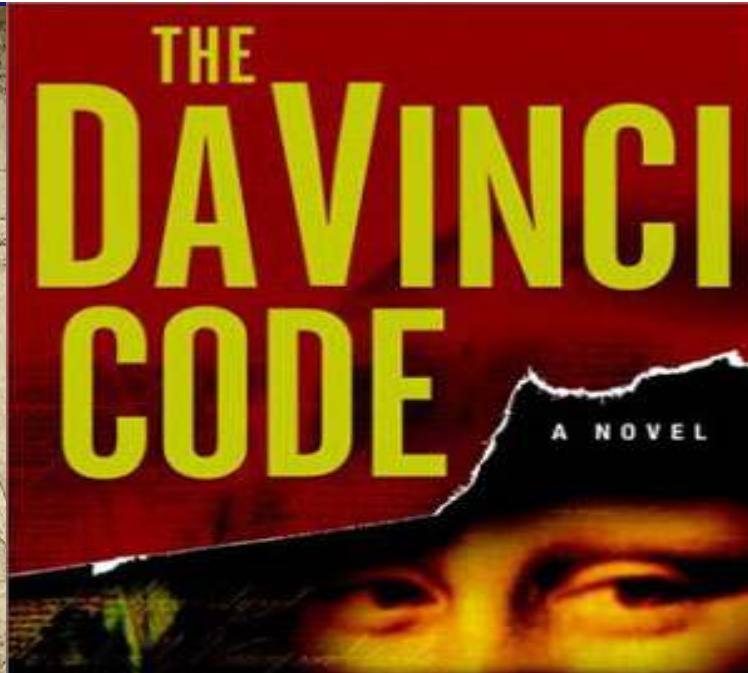
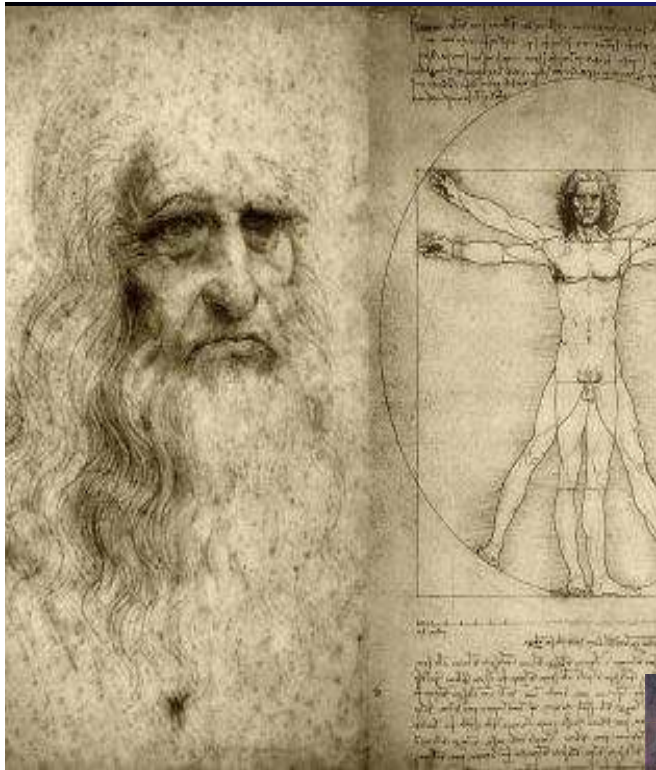
(always be alert to the presence of mixed disturbances)

### Correct disturbance

(buy time, don't over do)

### Correct disease

(a **MUST** if possible)



# Decoding an ABG Report

## Look at pH

- Low pH-----Acidosis
- High pH-----Alkalosis
- Normal pH--- A normal pH does not rule out existence of an acid base disorder (see below)

# If pH is abnormal Examine PCO<sub>2</sub> and HCO<sub>3</sub> relationship

- If PCO<sub>2</sub> and HCO<sub>3</sub> change in the same direction, it is a **single disorder**.

## Example:

- *in Metabolic Acidosis (i.e. decreased HCO<sub>3</sub>), PCO<sub>2</sub> should decrease. If it is decreased, consider Metabolic Acidosis with Respiratory Compensation (Single Disorder)*

## If pH is abnormal Examine PCO<sub>2</sub> and HCO<sub>3</sub> relationship (Cont)

- If PCO<sub>2</sub> and HCO<sub>3</sub> change in the opposite direction, it is a **double disorder**

### Example:

- *in Metabolic Acidosis (i.e. decreased HCO<sub>3</sub>), PCO<sub>2</sub> should decrease. If it is increased, consider Metabolic Acidosis with Respiratory Acidosis (Double Disorder)*

## Examine pH and HCO<sub>3</sub> relationship (For single disorders)

If pH and HCO<sub>3</sub> change in the same direction primary abnormality is **metabolic**

– Examples:

- *in metabolic acidosis both pH and HCO<sub>3</sub> decrease*
- *in metabolic alkalosis both pH and HCO<sub>3</sub> increase*

## Examine pH and HCO<sub>3</sub> relationship (For single disorders) (Cont)

If pH and HCO<sub>3</sub> change in the opposite direction primary abnormality is **respiratory**

- Examples:
  - *in respiratory acidosis pH decreases and HCO<sub>3</sub> increases*
  - *in respiratory alkalosis pH increases and HCO<sub>3</sub> decreases*



## If pH is normal Examine PCO<sub>2</sub> and HCO<sub>3</sub> relationship again

If PCO<sub>2</sub> and HCO<sub>3</sub> **grossly** change in the same direction, it is also a **double disorder**

### Examples:

- *in Metabolic Acidosis (i.e. decreased HCO<sub>3</sub>), PCO<sub>2</sub> should decrease to a certain extent. If it is decreased too much, consider Metabolic Acidosis with Respiratory Alkalosis (Double Disorder)*
- *in Respiratory Acidosis (i.e. increased PCO<sub>2</sub>), HCO<sub>3</sub> should increase to a certain extent. If it is increased too much, consider Respiratory Acidosis with Metabolic Alkalosis (Double Disorder)*

**If pH, PCO<sub>2</sub> and HCO<sub>3</sub> all are normal**

## **Normal Acid Base Status !!**

(----but please be careful of values on the extreme of reference ranges, which may be due to a fully compensated disorder like Chronic Respiratory Alkalosis or mild double disorders)

# ANION GAP

- It is a measure of anions other than  $\text{HCO}_3$  and Chloride
- Biochemical Basis:

Always:

$$\text{CATIONS} = \text{ANIONS}$$

# ANION GAP (Biochemical Basis)

- Major Cat-ions:

– Na = 140 mmol/L

– K = 4 mmol/L

– Ca = 4.5 mmol/L

– Mg = 1.5 mmol/L

TOTAL CATIONS = 150 mmol/L

# ANION GAP (Biochemical Basis)

- Major Anions:

– Cl	=	100 mmol/L
– HCO <sub>3</sub>	=	27 mmol/L
– proteins	=	15 mmol/L
– PO <sub>3</sub>	=	2 mmol/L
– SO <sub>3</sub>	=	1 mmol/L
– Organic Acids	=	5 mmol/L

TOTAL ANIONS = 150 mmol/L

# Anion Gap:

- Calculated as following:

$$(\text{Na} + \text{K}) - (\text{Cl} + \text{HCO}) = 18 \text{ mmol/L}$$

Range : 7 – 18 mmol/L

- It is only a lab derived index
- There can *Never* be an anion gap in any condition because electroneutrality is always maintained in the plasma

# High Anion Gap (Hyperkalaemic Metabolic Acidosis)

- Increased unmeasured Anions
  - Ketoacidosis
  - Lactic Acidosis
  - Renal Failure
  - Poisoning
    - Methanol
    - Ethanol
    - Salicylates
    - others

# Normal Anion Gap (Hypokalaemic Metabolic Acidosis)

- Diarrhoea:
  - Loss of  $\text{HCO}_3^-$
  - Loss of K
- Renal Tubular Acidosis
- Acetazolamide therapy
- Uretro-ileal shunt



# DELTA RATIO

The delta ratio is used for the determination of a mixed acid base disorder in an elevated anion gap metabolic acidosis

$$\frac{\text{Measured anion gap} - \text{Normal anion gap}}{\text{Normal [HCO}_3^-] - \text{Measured [HCO}_3^-]}}$$

or

$$\frac{(\text{anion gap} - 12)}{(24 - [\text{HCO}_3^-])}$$

# **Delta Ratio Assessment** **Guideline**

<b>&lt; 0.4</b>	Hyperchloraemic normal anion gap acidosis
<b>0.4 - 0.8</b>	<ul style="list-style-type: none"><li>• Renal failure</li><li>• Combined high AG &amp; normal AG acidosis</li></ul>
<b>1 to 2</b>	<ul style="list-style-type: none"><li>• Uncomplicated high-AG acidosis</li><li>• Lactic acidosis: 1.6 (average value)</li></ul>
<b>&gt;2</b>	A pre-existing elevated HCO <sub>3</sub> level due to: <ul style="list-style-type: none"><li>–a concurrent metabolic alkalosis, or</li><li>–a pre-existing compensated respiratory acidosis</li></ul>

# Predicted HCO<sub>3</sub>

$$\text{Predicted [HCO}_3\text{]} = 24 - [\text{Anion Gap} - 12]$$

- If patient's [HCO<sub>3</sub>] > Predicted [HCO<sub>3</sub>]
  - Metabolic acidosis + Metabolic Alkalosis
  - Compensatory response to resp acidosis

If patient's [HCO<sub>3</sub>] = Predicted [HCO<sub>3</sub>]

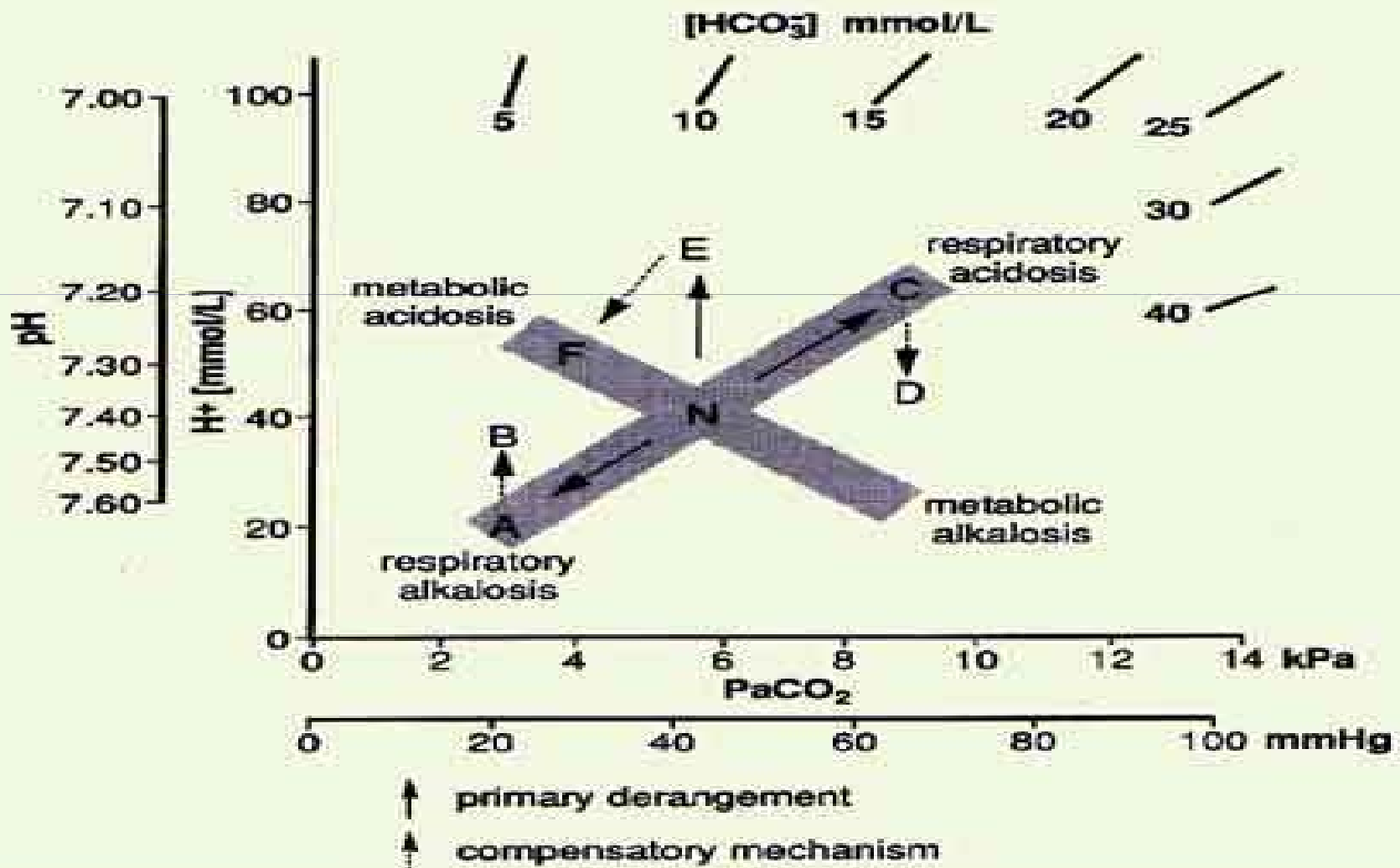
- Simple Metabolic acidosis
- If patient's [HCO<sub>3</sub>] < Predicted [HCO<sub>3</sub>]
  - Compensatory response to resp alkalosis

# Other Interpretation Assistants

- Flenley's Graph
- Electronic Calculators (*Skyscape* for desktop, ipads etc).
- Equations (or Formulas) – usually required for interpreting more subtle changes in ABGs.

# Flenley's graph

By plotting the result of the pH, HCO<sub>3</sub> and PaCO<sub>2</sub> on the graph, a mixed ABG disturbance is present if the point falls outside the arms of the CROSS.



# Learning Points

- **ABG samples should be sent to the lab immediately and ask for results instantly**
- **Mixed disturbances are very common in critically ill patients because of multiple pathologies**
- **Always be on the lookout for Mixed Disturbance**



# Acknowledgement

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